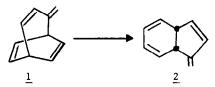
MECHANISTIC STUDY OF THE SINGLET EXCITED STATE PHOTOREARRANGEMENT OF 2-METHYLENE-BICYCLO[3.2.2]NONA-3,6,8-TRIENE. AN UNUSUAL [1,3] VINYL SHIFT. Zeev Goldschmidt^{*} and A. Worchel Department of Chemistry, University of Bar-Ilan, Ramat-Gan, Israel.

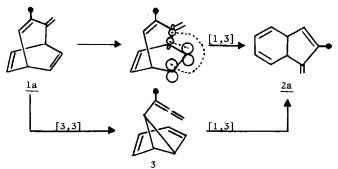
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In the preceding communication, we reported an interesting singlet state photochemical interconversion of methylenehomobarrelene $\underline{1}$ and dihydroindene $\underline{2}$.¹

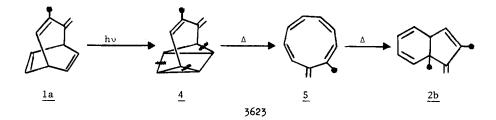


Two fundamentally different mechanisms for this interconversion seem to be possible <u>a priori</u>: Mechanism A whereby a <u>vinyl shift</u> takes place either directly as a photochemically allowed suprafacial [1,3] signatropic shift, or consecutively by a [3,3] signatropic shift to norcaradienyl allene <u>3</u> followed by a [1,3] shift to give <u>2a</u> (Scheme I), and Mechanism B where photochemical ${}_{\pi}2_{s} + {}_{\pi}2_{s}$ cycloaddition to form <u>4</u> is followed by thermal ${}_{2}2_{s} + {}_{\sigma}2_{s} + {}_{\sigma}2_{s}$ opening to nonafulvene (<u>5</u>), which recloses by a thermal disrotatory reaction to <u>2</u>.² (Scheme <u>II</u>).

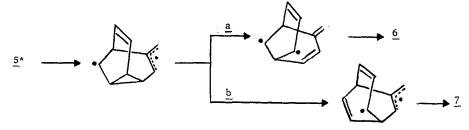
Scheme I. Mechanism A.



Scheme II. Mechanism B.



Formation of <u>6</u> and <u>7</u> by a sensitized irradiation of <u>5</u> represents another example of an unambiguous triplet excited state di- π -methane rearrangement of a "free rotor" bridged cycloheptadiene system.^{12,13} Furthermore, we note here an exceptional case of a non-regiospecific di- π methane rearrangement. Apparently, the two alternative pathways <u>a</u> and <u>b</u> (represented by Zimmermans qualitative valence bond mechanism), have similar excited state potential energy surfaces.¹⁴



Mechanistic aspects of the singlet rearrangement $5 \rightarrow 9$ are discussed in detail in the following communication.

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- 2) P. Ahlberg et al., ibid., 94, 7063 (1972).
- 3) J.C. Barborak, J. Daub, D.M. Follweiler and P.v.R. Schleyer, ibid., 91, 7760 (1969).
- 4) Rayonet RPR-100 reactor, "3000Å" lamps, pyrex filter.
- 5) Olefin 5: m/e 130; λ^C_{max}^L12 233 nm (ε 8500), 260 sh (5670); nmr (CDC1₃) δ3.38 (1H,q), 3.67(1H,t), 4.34 (1H,s), 5.01 (1H,d), 6.02 (1H,q), 6.20 (2H,t), 6.50 (2H,t). Cf. Ref.12. Satisfactory elemental analysis.
- 6) J. Grutzner and S. Winstein, <u>ibid.</u>, <u>94</u>, 2200 (1972). M.J. Goldstein and B.G. Odell, <u>ibid.</u>, <u>89</u> 6356 (1967).
- 7) Glpc analyses and separations were conducted using a 5% Carbowax 20M columns, at 90° .
- 8) Olefin <u>6</u>: m/e 130; nmr (CDCl₃) 61.68 (1H,m), 2.12 (1H,q), 2.34 (1H,d,q), 3.70 (1H,d,d), 4.82 (1H,s), 4.94 (1H,s), 5.40 (1H,d,d), 5.74 (2H,m), 6.34 (1H,d). Cf. Ref. 12.
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